PC3322

## **ONE HOUR THIRTY MINUTES**

A list of constants is enclosed.

## UNIVERSITY OF MANCHESTER

Nuclear Physics

28th May 2002, 9.45 a.m. - 11.15 a.m.

Answer <u>ALL</u> parts of question 1 and <u>TWO</u> other questions

Electronic calculators may be used, provided that they cannot store text.

The numbers are given as a guide to the relative weights of the different parts of each question.

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1. (a) The binding energy term of the semi-empirical mass formula may be expressed by:

$$B(A,Z) = a_v A - a_s A^{2/3} - a_c Z^2 A^{-1/3} - a_{sym} \frac{(A-2Z)^2}{A} \pm \delta.$$

Use this to explain why there is only one stable isotope in an isobar chain of a given odd-A. [5 marks]

- (b) Describe the main features of the nucleon-nucleon interaction. [5 marks]
- (c) State the angular momentum and parity selection rules for allowed *Fermi* and *Gamow-Teller* beta-decays. Classify the neutron beta-decay. [5 marks]
- (d) List *three* pieces of experimental evidence that supports the existence of a shell structure for the nucleus. [5 marks]
- (e) Explain why, in a prolate-deformed nucleus,  $\alpha$ -emission is more probable from the poles than from the equator. [5 marks]

2. Briefly describe the experimental method for measuring nuclear charge distributions via muonic X-ray spectroscopy. [6 marks]

Explain how muonic X-rays provide this information. [6 marks]

Compare the extent of the nuclear information that can be obtained from muonic X-rays with that from optical isotope shift measurements. [6 marks]

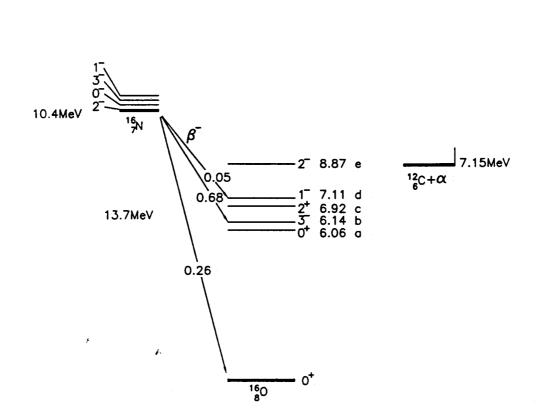
Estimate the energy of the muonic K X-ray  $(2p \rightarrow 1s)$  for  ${}^{56}_{26}$ Fe, assuming the nucleus is a point charge. Compare your answer with the experimental value of 1255 keV and comment on your result. [7 marks]

[Muon mass:  $m_{\mu} = 207 m_e$ .]

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13.7MeV

**3.** Given that the three lowest energy levels of a three-dimensional harmonic oscillator have orbital angular momentum values of  $\ell = 0$  (ground state),  $\ell = 1$  (at  $1\hbar\omega$ ) and  $\ell = 0$  and 2 (at  $2\hbar\omega$ ), sketch the single-particle levels arising from the 1s, 1p, 1d and 2s orbitals for a realistic nuclear potential and account for the level ordering. [9 marks]

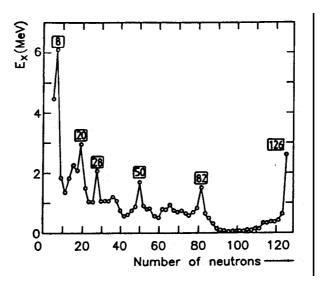


The diagram shows low-lying levels of <sup>16</sup>N and <sup>16</sup>O relative to the <sup>16</sup>O ground state. The threshold energy of <sup>12</sup>C+ $\alpha$  is also shown.

- (a) Give the most probable shell model description for the <sup>16</sup>N and <sup>16</sup>O ground states, and the state labelled b. [4 marks]
- (b) Explain why the <sup>16</sup>N ground state  $\beta^-$  decay to the <sup>16</sup>O ground state is less intense than its  $\beta^-$  decay to the 3<sup>-</sup> state. [5 marks]
- (c) Give the most probable decay modes of the states labelled a, b, c, d, giving the classification ( $\sigma$  L) of the gamma-ray transitions. [5 marks]
- (d) Explain why the state labelled e has no observed  $\alpha$  decay. [2 marks]

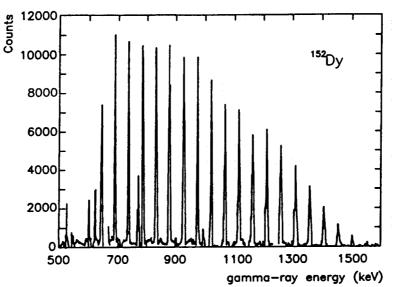
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4. The diagram shows the average energies of the lowest  $2^+$  states of even stable nuclei as a function of neutron number. Describe the characteristics of the level schemes you would expect to find in even stable nuclei in the regions (a) near N=64; (b) at N=82; (c) near N=100. [15 marks]



Show that, in a rotational band with constant moment of inertia  $\mathcal{I}$ , the transition energies increase linearly with nuclear spin. [4 marks]

The energy spectrum of  $\gamma$  rays emitted by a sequence of rotational states in <sup>152</sup>Dy is shown below. The  $\gamma$  ray transitions are  $I \to (I-2)$  electric quadrupole transitions but the nuclear spins of the individual levels are not known. By considering the difference in the  $\gamma$  ray energies for successive decays, determine a value for  $\hbar^2/2\mathcal{I}$  for the rotational band in <sup>152</sup>Dy. [3 marks]



Compare your value with the rigid-rotor value  $\hbar^2/2\mathcal{I}_{rigid} = 6$  keV appropriate for ground state deformations in the N=100 region, and comment on your answer.[3 marks]